

#### BIOENERGY | DOC2021 |

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## **4<sup>™</sup> DOCTORAL COLLOQUIUM BIOENERGY**

M. Sc. Christian Wondra

Determination of flammability limits and laminar flame velocity of biogenic synthesis gases

14<sup>™</sup> SEPTEMBER 2021, KARLSRUHE

**Overview on this presentation** 



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#### 1. Motivation

#### 2. The Project "KonditorGas"

#### **3.** Laminar flame velocity and flammability limit

#### 4. Construction of the test rig

#### **Project KonditorGas - Overview**



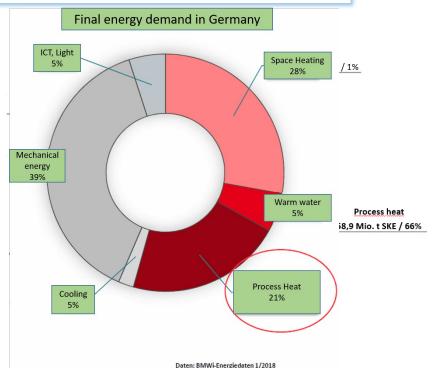
Title of the Doctoral Project:	"KonditorGas"				
Doctoral Student:	Christian Wondra				
Project Partners:	DBFZ Leipzig DBFZ E-Flox GmbH @eflox TesTneT GmbH				
University:	Friedrich-Alexander University Erlangen-Nürnberg				
University Supervisor:	Prof. DrIng. Jürgen Karl				
Funding :	Bundesministerium für Wirtschaft und Energie				
Duration:	09/2020 - 08/2023				

#### **Motivation – Industrial Process Heat**

#### III Handelsblatt

Industrie: Der heimliche Energiefresser: Grüne Lösungen für Prozesswärme gesucht

"Wenn wir wirklich wollen, dass die Energiewende funktioniert, müssen wir die Industrie auf CO2-neutrale Prozesswärme umstellen", sagt ... 12.08.2020



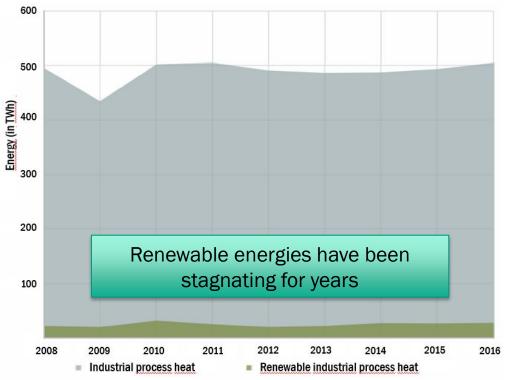
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- Process heat largest final energy consumer in industry
- 1/5 final energy consumption in
  Germany is required for process heat
- Renewable concepts important for achieving climate targets
- CO₂-tax since 2021
  → economic factor

### **Motivation – Energy sources for process heat**



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- Renewables contribute only 5 % to supply
- Main energy source is (natural) gas

- Substitution of natural gas by biogenic synthesis gases from the gasification of biomass
- Adaptation of the process chains for industry possible with little effort

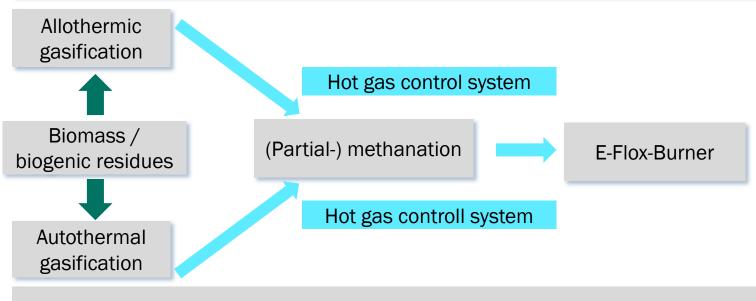
Hamburg Institut, Kurzgutachten zur Dekarbonisierung der Prozesswärme, 2018.

## Aim of the project: Complete process chain for direct utilization of biogenic synthesis gases



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Semi-industrial scale by means of a 100 kW Heat-Pipe-Reformer



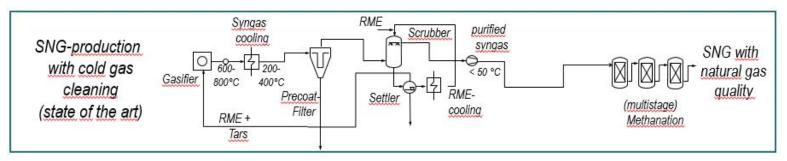
Proof-Of-Concept at a wood gasisfier site with real gas

# SNG production: State of the art vs. Catalytic conditioning

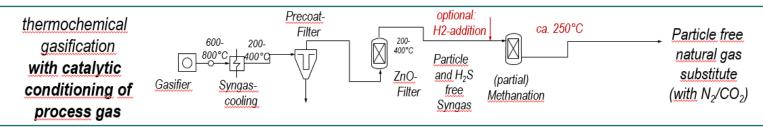


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#### State of the art: SNG with natural gas quality



Aim of the project: Simplification of gas purification and methanation, resulting in technical and economic simplification of the process chain



# Gasification: Different gas quality and inert gas content



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#### Allothermal process chain:

- 5 kW fluidized bed gasifier
- Steam as gasification medium

#### Autothermal process chain:

- 10 kW fixed bed gasifier (Spanner RE)
- > Air as gasification medium

$H_2$	CO	CO <sub>2</sub>	CH <sub>4</sub>	Tars
Vol %	Vol %	Vol %	Vol %	mg/m <sup>3</sup>
47,4	14,6	27,5	10,5	5000

H <sub>2</sub>	CO	CO <sub>2</sub>	CH <sub>4</sub>	N <sub>2</sub>	Tars
Vol %	Vol %	Vol %	Vol %	Vol %	mg/m <sup>3</sup>
19,5	22,6	11,1	1,9	43,7	193

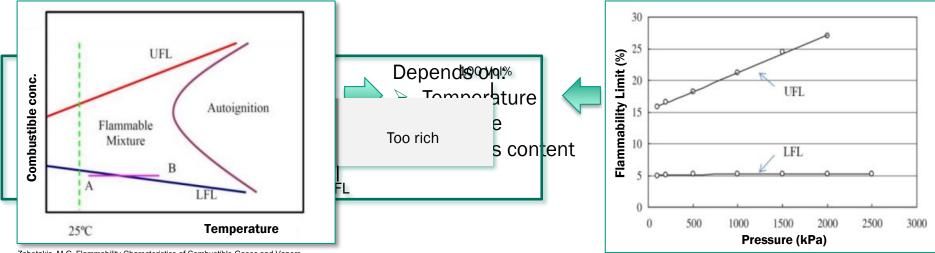


Due to the simplification of the process, the fuel gas contains higher concentrations of inert gases, which influence parameters such as flammability limits and flame velocities.

## Theoretical background – Flammability limit of gas mixtures



- Describes the flammability of a gas mixture in air
- Important safety characteristic
- Lower (LFL) and upper (UFL) flammability limit



Zabetakis, M.G, Flammability Characteristics of Combustible Gases and Vapors

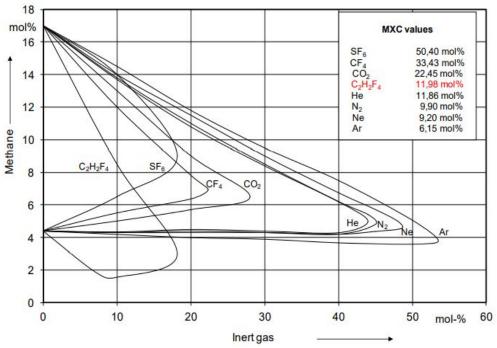
## Flamability limit – Influence of inert/neutral gas



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- Considerable effect on chemical reaction mechanism
- > Dilution of the fuel gas mixture
- Inert gas with high thermal capacity reduce the flame temperature
- Also the thermal conductivity is a factor

With increasing inert gas content the range between LFL and UFL is reduced



CHEMSAFE®, Database of evaluated safety characteristics

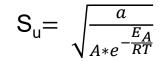
#### **Theoretical background – Laminar flame velocity**



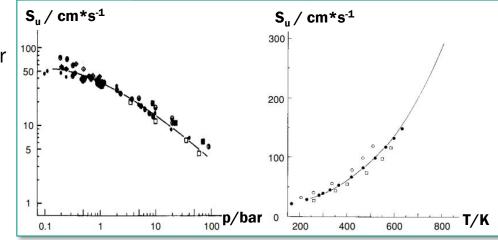
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- Corresponds to the propagation velocity of the flame front in the direction of the fuel/air mixture flowing after it
- Flame speed describes reactivity of the fuel gas
- Depends on the fuel/air mixture
- Defined by Zeldovich et al. as





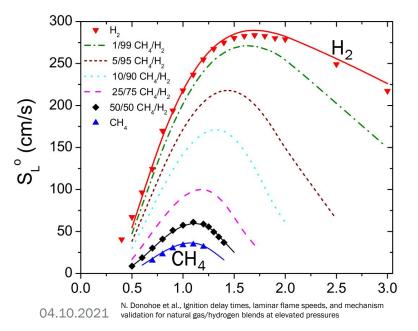


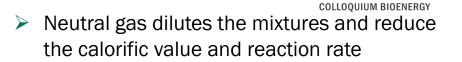
Warnatz et al., Combustion

> Depending on temperature and pressure

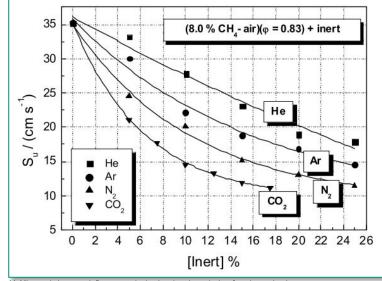
## Laminar flame velocity – Influence of gas composition

- H<sub>2</sub> has a considerably higher laminar flame velocity
- Depends on fuel-air equivalence ratio





Flame velocity decreases with higher neutral gas content



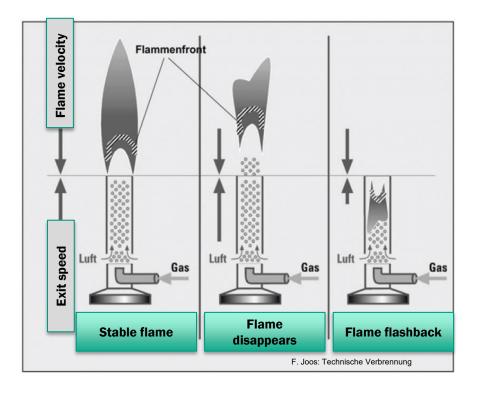
M. Mitu et al., Inert gas influence on the laminar burning velocity of methane-air mixtures

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# Laminar flame velocity – Important parameter for burner technology





- For a stable flame, the exit speed of the fuel/air mixture must be equal to the flame velocity
- Safety risks if the exit speed is too low



Flame flashback

Flame velocity too low



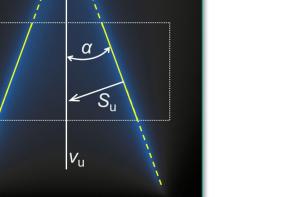
no continuous flame and bad performance of the burning system



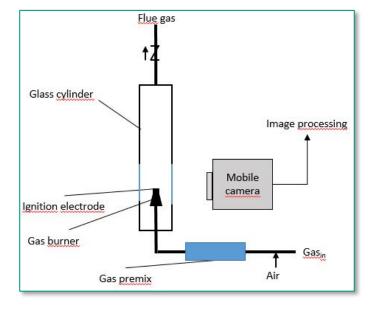
## Test rig – Determination of the laminar flame velocity

 Optical measuring methode via angle methode

 $S_u = v_u * \sin \alpha$ 



S. Richter, DLR-Institut für Verbrennungstechnik.



- Image processing with MatLab
- Mobile phone is used as a camera

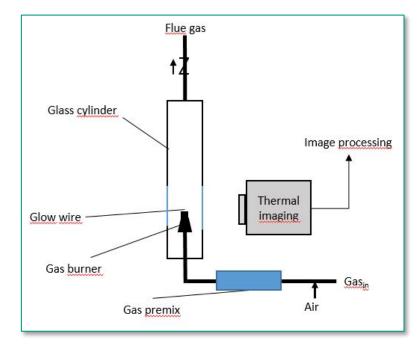
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## **Test rig – Determination of the flammability limit**



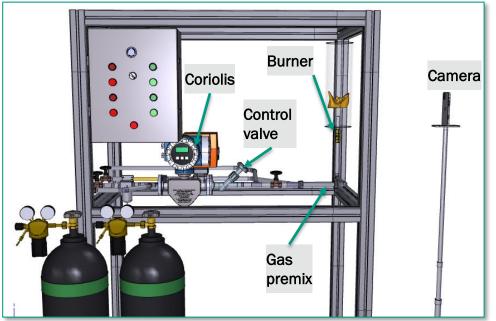
- Integration in the same test rig
- Acording to DIN Norm there are two different methods
- → Both methods are not continous
- > A modified tube methode is used
- Premixed gas flows over permanent ignition source
- → Flame formation or temperature changes are recorded
- → Validation of the new measurement methode with methane



### **Construction of the test rig**

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- The syngas volume flow is controlled by a coriolis sensor proportional valve combination
- The pipe section and measuring instruments are trace heated and insulated to prevent tar condensation (T > 150 ° C)
- For safety reasons N<sub>2</sub> can evacuate the whole test rig



### **Summary and outlook**

- Industrial process heat needs novel concepts to substitute fossil fuels
- Catalytically conditioned Syngas can be substitute natural gas and be used in modified burner systems
- The flame velocity is a decisive factor for the design and performance of the burner
- The higher inert gas and hydrogen content of the syngas can be influence the flame velocity as well as the flammability limits
- The test rig is used to determine the parameters for real wood gases and to investigate correlations with the performance of the burner







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#### Contact

Christian Wondra Chair for Energy Process Engineering FAU Erlangen-Nürnberg Fürther Str. 244f, D-90429 Nürnberg Phone: +49 911 5302 9399 Email: christian.wondra@fau.de

#### Karlsruher Institut für Technologie

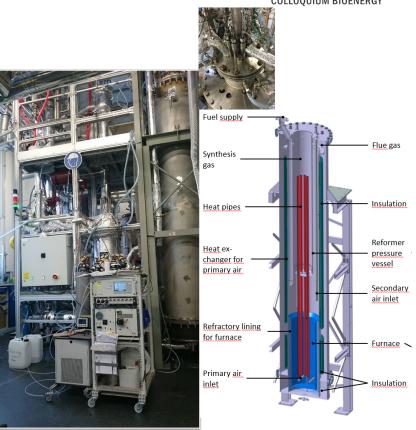
Kaiserstraße 12 D-76131 Karlsruhe Tel.: +49 721 608-0 Fax: +49 721 608-44290 **E-Mail: info@kit.edu** 



## **Proof of concept – Allothermic steam gasification**

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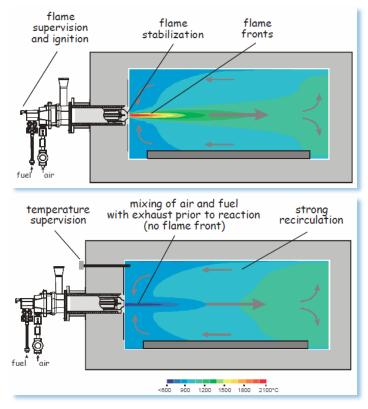
- > 100 kW Heat-Pipe-Reformer
- Construction of the whole process chain at EVT
- Heat input into the reformer through 8 high-temperature sodium heat pipes
- Demonstration on a semi-industrial scale



### **FLOX Technology – FlameLess OXidation**



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#### Advantages FLOX-Technology:

- 1. Oxidation in the combustion chamber volume instead of at the flame boundary
- 2. Uniform temperature distribution
- 3. Low thermal NOx
- 4. Excess air can be reduced
- 5. Fuel composition can vary
- 6. Weak gases can be used

#### Challenges of the FLOX-Technology:

- > Only works at high temperatures (>800  $^{\circ}$  C)
- Flame operation necessary when starting
- Until now natural gas is necessary for flame operation